

SHORT COMMUNICATION

ALKALOID BIOSYNTHESIS IN *ECHINOCEREUS MERKERI*

I J MCFARLANE and M. SLAYTOR

Department of Biochemistry, The University of Sydney, Sydney, N S W 2006, Australia

(Received 11 May 1971)

Abstract—The biosynthesis of 3,4-dimethoxyphenylethylamine and salsoline in *Echinocereus merkeri* has been studied using D,L-2'-¹⁴C-tyrosine, 1',2'-³H-dopamine Ring-(G)-³H-3-hydroxy-4-methoxyphenylethylamine and 1-methyl-1-¹⁴C-6-hydroxy-7-methoxy-1,2,3,4-tetrahydroisoquinoline-1-carboxylic acid. The first three were incorporated into 3,4-dimethoxyphenylethylamine but only the last two were incorporated into salsoline.

INTRODUCTION

THE ALKALOIDS produced by *Echinocereus merkeri* fall into three groups, tyramine derivatives, dopamine derivatives and the 1,2,3,4-tetrahydroisoquinoline, salsoline¹. This diversity must mean that their biosynthesis involves problems common to many more complex systems but the simple structures of the *E. merkeri* alkaloids makes this plant a good system for investigation. 2'-¹⁴C-tyrosine, 1',2'-³H-dopamine, ring-(G)-³H-3-hydroxy-4-methoxyphenylethylamine and 1-¹⁴C-1-methyl-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline-1-carboxylic acid were available, and the present paper gives the information obtained about alkaloid biosynthesis in *E. merkeri* using these substrates in classical feeding experiments.

RESULTS AND DISCUSSION

Table 1 provides a summary of the feeding experiments and Table 2 records the results of degradation experiments carried out to confirm the incorporation of precursors. The results of feeding and degradation experiments are consistent with the scheme presented in Fig. 1.

The essentials of this scheme are that tyramine (II) is a branch point for hordenine (III) and all the other alkaloids, and dopamine (IV) is a branch point for 3,4-dimethoxyphenylethylamine (VI) and salsoline (IX). Further, the methyltransferases necessary for the conversion of tyramine to hordenine and for the conversion of dopamine to 3-hydroxy-4-methoxyphenylethylamine (VII) appear to be blocked. The main conclusion from the feeding experiments is that alkaloid synthesis is directed to 3,4-dimethoxyphenylethylamine and not to salsoline or hordenine. The incorporation of tyrosine into tyramine and of tyrosine and dopamine into 3,4-dimethoxyphenylethylamine does not give any information on the role of dopa. Thus there remains the possibility that the reaction sequence tyrosine → dopa → dopamine is operating. The biosynthesis of hordenine (III) in barley from tyrosine and tyramine is well established.² In *E. merkeri*, tyrosine is efficiently incorporated into tyramine but not into hordenine. This can only be explained by the non-availability of the *N*-methylating enzymes at this stage of the plant's development. The non-incorporation of both

¹ S. AGURELL, J. LUNDSTROM and A. MASOUD, *J. Pharm. Sci.* **58**, 1413 (1969).

² E. LEETE, S. KIRKWOOD and L. MARION, *Can. J. Chem.* **30**, 749 (1952).

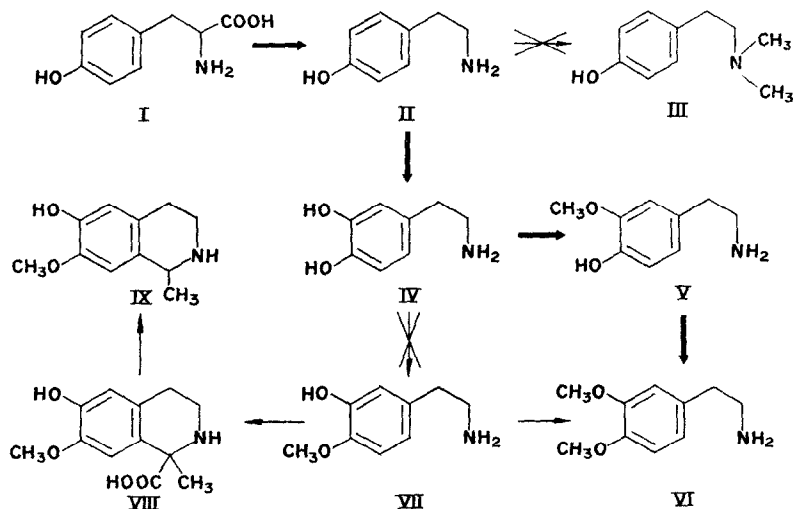


FIG. 1

tyrosine and dopamine into salsoline (IX) is surprising particularly as 3-hydroxy-4-methoxyphenylethylamine (VII) is incorporated. These results point to a blockage in the pathway, specifically to the methylating enzyme which converts dopamine into (VII). There are a number of possible intermediates between (VII) and salsoline which differ in the origin of the C-2 unit. Two of these possible intermediates are suitable for incorporation studies. One is the *N*-acetyl amine from (VII), namely *N*-acetyl-3-hydroxy-4-methoxyphenylethylamine which could be formed from acetyl CoA and (VII). This amide is not incorporated into salsoline in *E. merkeri*.³ A second possibility is 1-methyl-6-hydroxy-7-methoxy-1,2,3,4-tetrahydroisoquinoline-1-carboxylic acid, (VIII) which could be synthesized in the plant from (VII) and pyruvate. This acid is incorporated into salsoline in *E. merkeri*. An analogue of this acid, 1-methyl-6,7-dimethoxy-8-hydroxy-1,2,3,4-tetrahydroisoquinoline-1-carboxylic acid has been found⁴ in *Lophophora williamsii* and proposed

Table 1 FEEDING EXPERIMENTS TO *E. merkeri*

Substrate	Specific activity (mc/m-mole)	μ moles injected	μ moles ¹⁴ C-product formed	
			3,4-Dimethoxyphenylethylamine	Salsoline
D,L-2'- ¹⁴ C-Tyrosine*	50	0.74	0.004	0.000
1',2'- ³ H-Dopamine HCl	390	0.2	0.0015	0.000
Ring-(G)- ³ H-3-hydroxy-4-methoxyphenylethylamine HCl	13.6	9.34	0.84	0.28
1-Methyl-1- ¹⁴ C-6-hydroxy-7-methoxy-1,2,3,4-tetrahydroisoquinoline-1-carboxylic acid	0.068	4.4	—	0.006

* Tyrosine gave 0.036 μ moles of tyramine but no detectable amount of hordenine

³ I. J. McFARLANE and M. SLAYTOR, *Phytochem.* **11**, 229 (1972)

⁴ G. J. KAPADIA, M. B. E. FAYEZ, Y. S. VAISHNAV, H. M. FALES and G. SUBBA RAO, *Lloydia* **32**, 525 (1969)

TABLE 2 PERMANGANATE OXIDATION OF PHENYLETHYLAMINES FROM *E merkeri*

Substrate	Product	Specific activity (dpm/ μ mole)	3,4-Dimethoxybenzoic acid*	<i>p</i> -Methoxybenzoic acid*
D,L-2'- ¹⁴ C-Tyrosine	Tyramine HCl	119	—	0.8
D,L-2'- ¹⁴ C-Tyrosine	3,4-Dimethoxyphenylethylamine HCl	154	1.1	—
1',2'- ³ H-Dopamine HCl	3,4-Dimethoxyphenylethylamine HCl	314	1.8	—
Ring-(G)- ³ H-3-hydroxy-4-methoxyphenylethylamine HCl	3,4-Dimethoxyphenylethylamine HCl	6182	95.0	—

* Per cent of product activity

as an immediate precursor of anhalonidine. The major alkaloids (80%) in *E merkeri* are 3,4-dimethoxyphenylethylamine and its *N*-methyl derivatives¹ and it is perhaps not surprising that in these feeding experiments alkaloid synthesis is directed towards their production. That is, both tyrosine and dopamine are incorporated into 3,4-dimethoxyphenylethylamine (VI). In addition (VII) was also incorporated into (VI). It is suggested that (VII) is not a normal intermediate in the biosynthesis of (VI) and that (VI) is being produced from it by non-specific methylation. Rather, the true intermediate is 3-methoxy-4-hydroxyphenylethylamine (V) which has been found in *E merkeri*.¹ This means that the direction of methylation of dopamine determines whether the plant makes (VI) and its derivatives or whether it makes salsoline. It is interesting to note that in *Trichocereus pachanoi* which produces (VI) a detailed gas chromatography mass spectrometric analysis of the phenolic fraction could only detect (V) and not (VII).⁵ The importance of methylation in alkaloid synthesis in *L. williamsii* has been noted.⁶⁻⁸ The methylation of 3-methoxy-4,5-dihydroxyphenylethylamine apparently determines whether mescaline or the tetrahydroisoquinolines are synthesized. Thus 3-methoxy-4,5-dihydroxyphenylethylamine is a precursor for both alkaloids^{6,7} while the 3,4-dimethoxy-5-hydroxy derivative is an intermediate for the tetrahydroisoquinolines⁶ and the 3,5-dimethoxy-4-hydroxy derivatives is essentially only an intermediate in mescaline synthesis.⁸

The results illustrate the caution which must be used both with positive and negative results from feeding experiments. For example it is unlikely that tyrosine is not a precursor of all the alkaloids under discussion but reports of such non-incorporation of basic precursors are too rare in the literature.

EXPERIMENTAL

General. Plant material, chromatography, autoradiography, m.p. determinations and radioactive counting were as described previously.³

Administration of radioactive substrates. All radioactive substrates were injected in aq. solution. The tetrahydroisoquinoline carboxylic acid (VIII) was converted to the HCl-salt by cautious addition of 0.01 M HCl. After injection the cacti were grown for 3 weeks.

Isolation and detection of alkaloids. This was as described previously.³ In addition, tyramine HCl, hordenine sulphate and 3,4-dimethoxyphenylethylamine HCl were recrystallized from MeOH-EtOAc. They

¹ S. AGURELL, *Lloydia* **32**, 40 (1969).

² K. L. KHANNA, M. TAKIDO and H. ROSENBERG, *Phytochem.* **9**, 1811 (1970).

³ K. L. KHANNA, H. ROSENBERG and A. G. PAUL, *Chem. Commun.* 315 (1969).

⁴ A. G. PAUL, K. L. KHANNA, H. ROSENBERG and M. TAKIDO, *Chem. Commun.* 838 (1969).

were detected on TLC with 0.5% *o*-dianisidine in 1 M HCl and 10% NaNO₂ (tyramine and hordenine) or iodoplatinate (tyramine and 3,4-dimethoxyphenylethylamine)

Chemical syntheses 1-Methyl-1-¹⁴C-1,2,3,4-tetrahydroisoquinoline-1-carboxylic acid (VIII) This was synthesized by a modification of the method of Hahn and Rumpf⁹ Sodium 2-¹⁴C-pyruvate (0.05 mc, 82.5 mg) and 3-hydroxy-4-methoxyphenylethylamine HCl (102 mg) were dissolved in H₂O (0.5 ml) and the pH adjusted to 6 with dil. NH₄OH. Incubation for 4 days at 20–25° gave white crystals of (VIII). It was recrystallized from H₂O (100°) yield 53 mg (45%), m.p. 252° (reported 254°)⁹ Specific activity 0.068 mc/m-mole. For *in vivo* experiments (VIII) was further purified on TLC (neutral plates with solvent system A).

Ring-(G)-³H-3-hydroxy-4-methoxyphenylethylamine HCl Prepared from 3-hydroxy-4-methoxyphenylethylamine HCl and (CF₃)₂CO and ³H₂O. The exchange was effected by allowing the reactants to stand for 7 days at –15° in a sealed tube. M.p. 203–205° (reported 206–207°). Specific activity 13.6 mc/m-mole. Single spot on autoradiography of TLC (neutral plates with solvent system C). In a parallel experiment in D₂O it was shown by NMR that 31 per cent of the aromatic protons had exchanged.

O-Methylhordenine methiodide Tyramine HCl-ide (211 mg) was added to EtOH (10 ml) in which Na (100 mg) had been dissolved. The mixture was refluxed for 5 min in N₂. MeI (3 ml) was then added and the refluxing continued for 2 hr. NaHCO₃ (0.5 g) and more MeI (3 ml) were then added and the mixture was refluxed for 3 hr. The residue obtained on evaporation was extracted with hot CHCl₃ which was then filtered and evaporated. The residue was crystallized from EtOH affording colourless plates of *O*-methylhordenine methiodide m.p. 230–231° (reported 230–231°)¹⁰

Permanganate oxidation of phenylethylamines The phenylethylamine (or *o*-methylhordenine methiodide) (c. 100 mg) was dissolved in 5 ml of H₂O and NaOH (0.05 ml, 20%) KMnO₄ (16 ml, 3%) was added and the solution gently boiled under reflux for 30 min. While still boiling, excess KMnO₄ was destroyed by the addition of a few drops of EtOH. The solution was filtered, the filtrate acidified with HCl and extracted with CHCl₃ (4 ×). The combined CHCl₃ extracts were dried (CaSO₄) and the solvent removed under N₂. The crystalline residue was recrystallized 2 × boiling H₂O.

Acknowledgement—This work was supported by a grant from the Australian Research Grants Committee.

⁹ G. HAHN and F. RUMPF, *Chem. Ber.* **71**, 2141 (1938).

¹⁰ G. BARGER, *J. Chem. Soc.* **95**, 2197 (1909).

Key Word Index—*Echinocereus merkeri*, Cactaceae, alkaloids, biosynthesis, tyramines, isoquinolines